Small and stable peptidic PEGylated quantum dots to target polyhistidine-tagged proteins with controlled stoichiometry

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ABSTRACT

The use of the semiconductor quantum dots (QD) as biolabels for both ensemble and single molecule tracking requires the development of simple and versatile methods to target individual proteins in a controlled manner, ideally in living cells. To address this challenge, we have prepared small and stable QDs (QD-ND) using a surface coating based on a peptide sequence containing a tricysteine, poly (ethylene glycol) (PEG) and an aspartic acid ligand. These QDs, with a hydrodynamic diameter of 9 \pm 1.5 nm, can selectively bind to polyhistidine-tagged (histag) proteins in vitro or in living cells. We show that the small and monodisperse size of QD-ND allows for the formation of QD-ND-histag protein complexes of well-defined stoichiometry and that the 1:1 QD/protein complex can be isolated and purified by gel electrophoresis without any destabilization in the nanomolar concentration range. We also demonstrate that QD-ND can be used to specifically label a membrane receptor with an extracellular histag expressed in living HeLa cells. Here, cytotoxicity tests revealed that cell viability remains high under the conditions required for cellular labeling with QD-ND. Finally, we apply QD-ND complexed with histag end binding protein-1 (EB1), a microtubule associated protein, to single molecule tracking in *Xenopus* extracts. Specific co-localization of QD-ND-EB1 with microtubules during the mitotic spindle formation demonstrates that QD-ND and our labeling strategy provide an efficient approach to monitor the dynamic behavior of proteins involved in complex biological functions.

KEYWORDS (Word Style "BG_Keywords"). Quantum dot, peptide, Histidine-tagged Protein, stoichiometry, End Binding 1 protein, microtubule

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INTRODUCTION

The development of simple coupling methods to visualize, track and eventually activate proteins using fluorescent labels in living cells is a key step in cellular imaging. ¹⁻⁶ Although fluorescent proteins have revolutionized this field, they have shortcomings, such as rapid photobleaching of the chromophores and the requirement of genetically-modified proteins, which have necessitated the development of alternative approaches to chemically label proteins in live cells.^{3, 7} In this view, fluorescent semiconductor nanocrystal quantum dots (QDs) have received a considerable attention because of their unique optical properties, which are well suited for biological imaging.⁸⁻¹⁰ Many strategies have now been described to obtain bioconjugated QD probes, including both covalent or non-covalent interactions of proteins or other biomolecules at the QD surface. These include specific recognition system such as streptavidin/biotin,^{7, 11, 12} antibody-antigen recognition¹³ or electrostatic interactions.¹⁴ General concerns in all these methods are the final size of the bioconjugate 15, 16, the multivalency of the conjugation and the preservation of protein functionality. ¹⁷ Since the size of a QD is comparable or greater than the size of many proteins, it is technically difficult to separate QD-protein complexes with different stoichiometries. Only recently, it was reported that gel electrophoresis can be used to isolate QD-streptavidin/antibody complexes with 1:1 stoichiometry.¹⁸

To target a specific site on a protein of interest, one strategy is to introduce a small fusion tag¹⁹ such as a polyhistidine peptide sequence (histag). This tag, commonly used for purification, contains six (H6) or ten histidines (H10) and exhibits a high affinity for divalent nickel (Ni²⁺) or zinc (Zn²⁺) ions. Previously, we reported the preparation of QD-micelles bearing a trisnitrilotriacetic acid (trisNTA) ligand to target histag proteins by high-affinity complexation via Ni²⁺ ions. ²⁰ However, a possible limitation of these OD-micelles is that

their hydrodynamic diameter is between 20 to 30 nm following complexation with the histag protein. Since intracellular trafficking of nanoparticles²¹ and their access to crowded cellular environments²² are strongly influenced by their size, it is important to develop strategies in order to reduce the overall size of QD-protein complexes. One effective route has been the use of compact surface coatings which enable the direct binding of histag proteins to exposed or accessible Zn²⁺ ions at the surface of CdSe/ZnS QDs.²³⁻²⁶ For this purpose, PEGylated dihydrolipoic acid (DHLA) have been shown to facilitate the direct interaction of histag proteins with the QD surface.^{18, 27,28} Using the DHLA strategy, QD-protein complexes exhibit a reduced size (hydrodynamic diameter ~ 15 nm) in comparison to QDs with amphiphilic molecule (~20-30 nm) ^{29,30,31} or polymer (~20-30 nm)³² based surface coatings.

In the present work, we report a simple alternative for labeling individual histag proteins with QDs while maintaining a small QD/protein complex size. We selected short peptides to stabilize the QDs in aqueous buffers because of their good chemical versatility as well as their higher chemical stability compared with QD coatings formed with ester bonds.^{33, 34} Peptides also have the advantage that it is easy to introduce biologically active sequences as terminal groups and they can be obtained from commercial providers, eliminating the need for complex or time-consuming chemical synthesis and characterization. Peptides derived from the phytochelatin were first demonstrated to be a very effective surface coating, yielding water-soluble QDs with hydrodynamic diameter of ~10-15 nm.³⁵ We have found that small tricysteine peptide derivatives, such as the sequence CCCSSSD, can also interact strongly with the QD surface and serve to form stable water-soluble QDs with a hydrodynamic diameter of ~10 nm.³⁶ Based on these results, we prepared small QDs based on peptide ND, composed of a tricysteine adhesive group, a hexamer of PEG and an aspartic acid grafted to the terminal group. We show that for these QD-ND, the PEG spacer effectively prevents non-specific adsorption on cell membranes and that the probe is nontoxic to cells under standard

experimental conditions. Furthermore, specific binding of histag proteins is mediated by binding to Zn^{2+} ions directly accessible on the QD surface and possibly enhanced by the metal chelating dicarboxylate acid functionality of the aspartic acid.

Through application of QD-ND, we demonstrate the ability to label and track histag proteins both in vitro and in living cells. In vitro, we show evidence from dynamic light scattering (DLS) and fluorescence correlation spectroscopy (FCS) measurements that the OD-ND/protein complex is stable, small and can be prepared with 1:1 stoichiometry following purification by gel electrophoresis. To show specific binding in live cells, we targeted a membrane receptor with an extracellular polyhistidine sequence expressed in HeLa cells. Finally, we applied our labeling strategy to record the movement of end binding protein-1 (EB1), a microtubule associated proteins in Xenopus cell extracts. EB1 is 35 kDa cytoplasmic protein known to specifically bind to microtubules and to regulate their dynamics.³⁷ In this study, a purified histag EB1 was coupled to QD-ND in vitro and applied to cell extracts. We monitored QD-ND-EB1 during the formation of the mitotic spindle, a microtubule-based structure essential for cell division.³⁸ and found that QD-ND-EB1 co-localized specifically with the spindle. Single molecule tracking allowed us to extract the velocity distribution of EB1 interacting with the microtubules. Variability observed in EB1 motion might reflect the various modes of the dynamics of the EB1-microtubule complex, illustrating the intrinsic dynamical heterogeneity of the mitotic spindle apparatus.

EXPERIMENTAL METHODS

Synthesis of the peptide ligands (see Figure 1 for reaction scheme):

All the reactants and solvents were purchased from Sigma Aldrich.

Hexaethyleneglycol mono-p-toluenesulfonate 2: Potassium iodide KI (0.295 g, 1.78 mmol, 0.1 eq), iron oxide Ag₂0 (4.95g, 21.4 mmol, 1.2 eq) and tosyl chloride were added to a

solution of commercial hexaethylene glycol (5.01 g, 17.8 mmol, 1 eq) in anhydrous CH₂Cl₂ (178 mL). The reaction mixture was kept at room temperature for 18 h and then filtrated on celite. After solvent evaporation, the crude product was subjected to chromatography (silica gel 60 g, elution with AcOEt/MeOH 85/15) to give the monotosylate **2** (7.25g, 16.6 mmol, yield 94 %). ¹H-NMR (200 MHz, CDCl₃):δ= 7.80 (d, J=8Hz, 2H), 7.34 (d, J=8Hz, 2H), 4.18 (t, 2H), 3.75-3.58 (m, 22H), 2.61(broad s, OH), 2.48 (s, 3H).

Hexaethyleneglycol monoazide **3**: **2** (7.25g, 16.6 mmol, 1 eq) and sodium azide were mixed in a solution of anhydrous DMF (36 mL). The mixture was kept at reflux for 22 hours under argon. After DMF evaporation, the mixture was extracted with diethylether and the organic phases were washed with brine. Drying over Na₂SO₄ afforded **3** (4.653 g, yield 90%). H-NMR (200 MHz, CDCl₃): δ = 3.72-3.67 (m, 22H), 3.41 (t, 2H), 2.78 (broad s, OH).

Tert-butyl 2-[Monoazide (Hexaethyleneoxy)] acetate **4**: Tertbutyl bromo acetate (4.68g, 24 mmol, 1.6 eq), iron oxide (0.49g, 0.2 eq) potassium iodide (5.21g, 1.5 eq) were added to a solution of **3** (4.653g, 15 mmol, 1 eq) in anhydrous CH_2Cl_2 (50 mL). The mixture was kept at room temperature for 5 days. After filtration on celite and solvent evaporation, the crude product was subjected to chromatography (silica gel, eluent AcOEt/MeOH 90/10) to afford **4** (6.089g,14.5 mmol, yield 96%). H-NMR (200 MHz, CDCl₃): δ = 4.05 (s, 2H), 3.73 -3.69 (m, 22H), 3.41 (t, 2H), 1.51 (s, 9H).

Tert-butyl 2-[amino (Hexaethyleneoxy)] acetate **5**: A suspension of **4** (6 g, 14.5 mmol, 1eq) and 0.6 g of Pd/C (10 % in mass) in anhydrous CH_2Cl_2 (60mL) was stirred for 45h under reduced pressure of dihydrogen. Filtration and solvent evaporation afforded **5** (5,09g, 12.9 mmol, yield 90%) that was used without any further purification. 1 H-NMR (300 MHz, CDCl₃: δ = 4.05 (s, 2H), 3.73 -3.69 (m, 22H), 3.95 (t, 2H),3.22 (t, 2H), 1.50 (s, 9H).

Tert-butyl 2-[Fmoc-amino (Hexaethyleneoxy)] acetate **6**: N-(9-fluorenylmethoxycarbonyloxy)succinimide (Fmoc-OSu) (1.97g, 5.8 mmol, 2.33 eq) was

added to a solution of **5** (0.98g, 2.48 mmol, 1 eq) in THF (8.6 mL) at 0°C. The reaction mixture was kept at room temperature for 42 h. The crude product was subjected successively to two chromatographies (silica gel 90 g, eluent AcOEt/ hexane (50/50 and 75/25)) to give **6** (0.85g, 1.82 mmol, yield 73%). The product **6** was then used as a classical protected Fmoc peptide. 1 H-NMR (300 MHz, CDCl₃: δ = 7.8 (d, 2H), 7.64 (d, 2H), 7.45 (t, 2H), 7.31 (t, 2H), 5.58 (t, 1H), 4.45 (d, 2H), 4.25 (t, 1H), 4.05 (s, 2H), 3.72 -3.66 (m, 22H), 3.45 (q, 2H), 1.50 (s, 9H).

Peptide ND synthesis: Peptides were synthesized by classical Fmoc (N-[9-fluorenyl] methoxycarbonyl) solid-phase chemistry and by using a commercial automatic peptide synthesizer by coupling Fmoc-α-amino acids on preloaded Wang resin. ^{36, 39} Protected amino acids and 6 were coupled by in situ activation with 2-(1H-Benzotriazole-1-yl)-1,1,3,3tetramethyluronium tetrafluoroborate (TBTU) and N-hydroxybenzotriazole (HOBt). NII-Fmoc deprotection was performed with 20% piperidine in DMF. Side chain deprotection and cleavage of peptides from the solid support was performed by treatment with cleavage reagent (94.5 % trifluoroacetic acid (TFA) / 2.5% water/ 2.5% ethanedithiol (EDT)/ 1%triisopropylsilane (TIS)) for 2h at 20°C. Peptides were purified by reversed-phase HPLC (RP-HPLC) using a Waters semi-preparative HPLC system on a X Terra 10µm column (300x19mm). The elution was performed with a gradient of aqueous 0.1% TFA (A) and 0.08% TFA in acetonitrile (B) at a flow rate of 10 mL/min with a photodiode array detection at 210-440nm. The purity of each peptide was controlled by analytical RP-HPLC on the same instrument with a X Terra 5µm column (250x4.6mm) using a linear gradient of 0.1% TFA in water and acetonitrile containing 0.08% TFA at a flow rate of 1 mL/min. Finally, the peptide was analyzed by electrospray spectrometry. Peptide ND: Purity 85 %, $t_r = 12.48$ min. HRMS $[M + H]^+$ calcd for $C_{27}H_{45}N_5O_{14}S_3$ 764.25164, found 764.2523 (1 ppm).

Peptide 7: The control peptide 7 (H_2N -CCCSSSD-OH) was prepared according our previous described method³⁶: Purity 85 % t_r = 12.48 min. HRMS [M + H] + calcd for $C_{22}H_{37}N_7O_{13}S_3$ 704.16898, found 704.1688 (1 ppm).

Peptide functionalization of QDs: Commercial core/shell CdSe/ZnS QDs (Evidots, Evident Technology, λ_{em} = 545 nm or 605 nm) were precipitated from initial toluene solution with methanol and dried under vacuum and then redispersed in a pyridine solution. Peptides ND or 7 were dissolved in anhydrous 99.8% dimethyl sulfoxide (DMSO), at a concentration between 10 and 30 mM. The QDs in pyrdine and the peptide solution in DMSO were mixed together. After addition of tetramethylammonium hydroxide, aggregates of fluorescent QD-peptide were isolated by centrifugation during 30 min at 15000 g. After removing the supernatant, the QDs were dried for 2 hours and then redispersed in water. To remove the aggregated nanocrystals, this solution was ultracentrifuged at 100 000 g during 30 min on a 50% sucrose cushion. The resulting sample was passed through a G25 spin column and dialyzed against pure water to remove sucrose and the excess of free peptide. Before using, the QD-peptide solution was filtered through a 0.22 μm Millipore filter (Millex HV, Sigma-Aldrich).

EB1 expression and purification: Full-length H6-EB1 was expressed in BL21-PlysS cells (Novagen) and purified on nickel-agarose beads. ⁴⁰ After dialysis against a sodium phosphate buffer, the proteins were flash frozen and stored at -80°C in aliquots.

Dynamic light scattering (DLS) and fluorescence correlation spectroscopy (FCS): The mean hydrodynamic diameters were obtained from a Nanosizer ZEN3600 (Malvern Instruments) DLS spectrometer (angle 90° at 25 °C). All samples were filtered through 0.22 μm Millipore filter. FCS measurements were performed on a home-built microscope equipped with an

Olympus 60X water immersion objective (UPlan Apo, NA 1.2). Two-photon excitation (200 fs, 76 kHz, 750 nm excitation) was provided by a mode-locked Ti-Sapphire laser (Mira, Coherent). The incident power at the sample was measured with a Lasermate powermeter (Coherent). The fluorescence photons were collected through filters (AHF Analysentechnik) and optical fibers (FG200LCR multimode fiber, Thorlabs) and detected with avalanche photodiodes (SPCM-AQR-14, Perkin Elmer) coupled to an ALV-6000 correlator (ALV GmbH). The geometrical characteristics of the focal point were determined using a fluorescein solution of known concentration (50 nM in 0.1 M NaOH). FCS measurements were performed on aqueous QD solutions in the nanomolar range at 20°C. Autocorrelation curves were well adjusted assuming a single freely diffusing species in solution. In the case of two-photon excitation, the diffusion time \mathbb{I}_D through the beam waist of size \mathbb{I} can be linked to the hydrodynamic radius R of the diffusing species using the Stokes-Einstein relation by: $\mathbb{I}_D = 3\pi\mathbb{I}\mathbb{R}\mathbb{I}^2/4kT$ where \mathbb{I} is the viscosity, T the temperature and k the Boltzmann constant.

Electrophoresis and protein coupling: QD-ND (200nM) was pre-loaded with Ni²⁺ (50μM) for 15 min, and then incubated with various molar ratio of histag EB1 during 1h. Gel electrophoresis of the different QD-ND-EB1 ratios was performed on a 0.5% agarose gel in in HEPES buffer (pH 7.5) under an applied voltage of 100V for 30 min.

Photophysical properties: UV-Vis spectra of QDs were performed on a Cary 100 UV-vis spectrophotometer (Varian). The QD diameters were evaluated from the maximum absorbance of the excitonic peak and the extinction coefficient at 350 nm was calculated according to the literature (for QDs with emission at 605 nm, ε = 10⁵ cm⁻¹.mol⁻¹). Fluorescence spectra were measured on a Spectrolog 3 (Horiba Jobin Yvon) fluorimeter.^{41, 42}

Quartz Crystal Microbalance: The quartz crystal microbalance (QCM-D) experiments were performed on a Q-Sense D 300 apparatus (Q-Sense AB) by monitoring the resonance frequencies of the gold coated crystal at four different frequencies (5, 15, 25 and 35 MHz). This data gives, in principle, information about the adsorption process as well as the mechanical properties of the deposited film. At the start, the crystal was put into contact with a buffer solution (Tris-Borate-NaCl: TrisBase 0.89 M, Boric Acid 0.89 M, NaCl 0.20 M) for 10 min. Next, a 0.5 mg/mL peptide ND solution was injected and left 50 min in contact with the sensor before three successive rinsing steps with the buffer. A solution of NiCl₂ (1.5 mM) was then injected followed by EB1 solution (0.07 mg/mL in buffer solution). After stabilization of the signal and a rinsing step with the buffer, EDTA solution (2 mM) was injected.

Cellular Cytotoxicity MTT assay: HeLa cell culture was performed in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% of fetal calf serum, 100 U/mL penicillin, 100 μg/mL streptomycin at 37 °C/5% CO₂. The HeLa cells were seeded in 96 multiwell plates (500 cells per well, 100 μL) 24 hours before adding the QDs. Then after growth medium removal, appropriate concentrations of QD-ND (filtered 0.22 μm) were added to each well. The typical protocol for MTT assay was as follows: A 5 mg/mL aqueous solution of MTT (Methylthiazolyldiphenyl-tetrazolium bromide, Sigma Aldrich) was added to HeLa cells (10 μL). After 4 hours of incubation at 37 °C/5% CO₂, the plates were centrifuged (5 min, 1000 rpm). After medium removal and addition of DMSO (150 μL), the plates were stirred until complete dissolution of the purple formazan crystals. Finally the absorbance at 540 nm was measured by a UV-Vis plate reader (Labsystems Multiscan RC) driven by Biolise software. The absorbance was normalized using absorbance of control cells grown in non-modified medium.

HeLa cell labeling: HeLa cells were transfected 24 hours before labeling with the gene expressing the membrane protein H10-CFP-TM, a transmembrane domain of platelet-derived growth factor receptor ⁷ as previously reported.²⁰ Cell labeling was performed by placing coverslips on a heating block at 37°C. Cells were labeled with 20 nM QD-ND by incubation in PBS for 10 minutes followed by repeated washing with PBS. HeLa cells were imaged at room temperature in OptiMem on an inverted wide-field epifluorescence microscope (Olympus, IX70) equipped with a 60X oil objective (NA 1.45, Olympus). Fluorophores were excited with a broad-spectrum mercury lamp source. Images were acquired using a back-illuminated CCD camera (Micromax 512EBFT, Roper Scientific) and MetaView solftware (Universal Imaging). H10-CFP-TM was imaged through a CFP filter set (excitation: 440±10 nm, dichroic: 455 nm, emission: 480±20 nm). QD-ND was imaged through a custom filter set (excitation: 420±10 nm, dichroic: 470 nm, emission: 595±10 nm).

QD monitoring of microtubules self-organization and spindle formation: Cytostatic-factor-arrested Xenopus laevis egg extracts were prepared as previously described.³⁸ Tubulins, labeled with X-Rhodamine (Molecular Probes) were prepared as previously described.³⁹ The molar ratio between QD-ND and EB1 was 1:6. A solution of QD-ND-EB1 complex solution in the nanomolar range was incubated 15 min after the beginning of the spindle assembly process. Fluorescence imaging of microtubule spindles was performed using an IX81 (Olympus) equipped with either a X60 (Plan Apo, NA 1.42) or a X100 (Plan Apo, NA 1.4) objective, and an EM-CCD (C9102, Hamamatsu). Image analyses were performed using Image J Software (Scion Image) and Simple PCI software. The mean velocity was calculated from the image sequence of the individual QD-ND-EB1 collected during 20 sec (N=87).

events). The motion of individual QDs was acquired with an integration time of 100-400 ms per frame.

RESULTS AND DISCUSSION

For the purpose of stabilizing QDs in aqueous buffer and enabling the conjugation of histag proteins, the short PEGylated peptide ND was synthesized according to the scheme presented in Figure 1. At the N-terminal, a tricysteine serves as an adhesive sequence designed to bind to the QD surface. An aspartic acid, which is negatively charged at basic pH, was grafted at the C-terminal to provide charge stabilization of the colloidal QDs in a variety of common buffers and at different pHs (see Figure S1 in supporting information). In addition, the aspartic acid was selected because of its expected ability to chelate metal ions and enhance the binding of histag proteins. Between the tricysteine and aspartic acid, a hexamer of PEG was introduced as a spacer to reduce non-specific interactions and improve solubility. To obtain this compound, the PEGylated amino acid 6 was synthesized in several steps. Hexaethylene glycol 1 was first activated as the monotosylate 2 in presence of an iron oxide catalysis and then converted into hexaethyleneoxy mono azide 3. After substitution of the free terminal alcohol with the tertbutyl bromoacetate and azide reduction of 4, the bifunctional amino and protected carboxylic acid derivative 5 was obtained. The amino group was then protected with Fmoc succinimide to give 6. This PEGylated amino acid 6 was first coupled to an aspartic acid pre-loaded Wang resin according to classical Fmoc (N-[9-fluorenyl] methoxycarbonyl) solid-phase chemistry. Three cysteines were successively coupled by using a HBTU/HOBt activation. Finally after cleavage and deprotection, the peptide ND was obtained. To form a control peptide without the PEG spacer (peptide 7), the compound 6 was replaced by triserine spacer (Figure 2a).

To test the ability of the terminal aspartic acid in peptide ND to bind histag proteins via Ni²⁺ chelation, a surface-based assay was performed using a quartz crystal microbalance (QCM) and the protein histag EB1. The peptide ND was deposited on the gold-coated quartz microbalance via the strong interaction between cysteines and the gold surface. The adsorption and desorption of the histag EB1 proteins to the immobilized ND peptide in the presence or absence of Ni²⁺ ions was monitored (Figure S2 and Table S1 in supporting information). After preloading the ND peptide surface with Ni²⁺ ions, adsorption of the histag EB1 was enhanced by more than 3-fold compared to in the absence Ni²⁺ ions (Table S1 in supporting information). The addition of an EDTA competitor induced a desorption of about 46% of the adsorbed protein, indicating that a large fraction of the histag EB1 was bound through the Ni²⁺ loaded aspartic acid of the ND peptide.

Commercial semiconductor CdSe/ZnS QDs (λ_{em} = 605 nm) were functionalized with the peptide ND or 7 according to a previously described method ³⁶ (see Figure **2a**) based on ligand exchange between the initial trioctylphosphine oxide ligands of the hydrophobic QDs and the peptides. ³⁵ Water-soluble QD-ND exhibited a hydrodynamic diameter of 9 nm (+/-1.5 nm), determined by DLS and FCS measurements (see Table **1**) and presented a high colloidal stability (over several months) in common buffers. During gel electrophoresis, the QDs migrated in a narrow band towards the positive electrode, which is consistent with their expected net negative charge (Figure **2b**). The optical properties of the QDs were not altered after functionalization with the peptides, as shown from absorbance and fluorescence spectra (Figure **2b**). ³⁶ In addition, the emission intensity was found to be stable between 6 and 12 and weakly affected until pH 5 (see Figure **S1** in supporting information). A strong quenching was then observed for acidic pHs (lower than 5). Therefore the QD-ND were found to be stable between pH 5 and 12. The effect of loading QD-ND with 50 μ M Ni²⁺ was also minimal (Figure **2b**), after optimizing the concentration and type of divalent cation (see Figure **S3** in

supporting information). The measured fluorescence quantum yield of the QD-ND was found to be around 25% in water, which is an improvement over other peptide-based surface coatings.³⁵

Figure 2

Next, we tested the effect of the PEG spacer on non-specific cellular interactions of the peptide stabilized QDs. We monitored the adhesion of PEGylated QD-ND and non-PEGylated QD-7 towards living HeLa cells following incubation for 10 minutes and analysis by fluorescence microscopy (Figure 3). Non-specific adsorption of the QD-7 on cell surface was present in high levels, indicated by the presence of QD fluorescence on the cell membranes, but was nearly completely inhibited by replacing the triserine spacer in 7 by the hexamer of PEG (case of QD-ND). Consequently, all the following experiments were performed with QD-ND.

Figure 3

The affinity of the QD-ND for histag proteins was then investigated in solution by following the interaction with DLS and FCS (Figure **S4** in supporting information). A systematic increase in hydrodynamic diameter was observed for QD-ND, following incubation with increasing amounts of histag EB1 (Table **1**). Considering the respective geometry of QD-ND and the EB1, the 10 nm increase in diameter observed for high ratios of the protein is in good agreement with the binding and steric saturation of the EB1 on the QD surface.

Mean hydrodynamic diameters obtained by	0 nEB1/nQD	1 nEB1/nQD	10 nEB1/nQD
FCS	13± 2 nm	16± 2 nm	24± 3 nm
DLS	9 ± 1.5 nm	14 ± 2.5 nm	$20 \pm 4 \text{ nm}$

Table 1

To provide further evidence of the conjugation between the QD-ND and the histag EB1, a titration of QD-ND with the protein was followed by gel electrophoresis. The addition of histag EB1 at various ratios to the QD-ND samples resulted in a systematic change in the QD migration patterns observed following electrophoresis (Figure 4a and Figure S5 in supporting information). QD-ND without EB1 migrated in a narrow, single band toward to positive electrode. However, in the QD-ND samples incubated with low ratios of EB1, discrete bands formed in the gel corresponding to different stoichiometries of the QD-ND-EB1 complex (Figure 4a). At higher ratios of the protein, the sample merged again to a single band because of the steric saturation of the QD surface with EB1.³⁰ Based on the ability to resolve these discrete bands in the gel, QD-ND containing exactly 1, 2 and 3 copies of EB1 were isolated and purified by gel extraction. The temporal stability of the isolated complexes was high, because re-analysis of the purified 1:1 QD-ND-EB1 after several hours of storage showed that the complex maintained the expected position in the gel with respect to the unconjugated QD-ND (Figure 4b). Thus, small and stable QD-protein complexes with defined stoichiometry could be prepared based on QD-ND.

In control experiments using gel electrophoresis, it was found that binding occurs between histag EB1 proteins and QD-ND without Ni²⁺ pre-loading of QD-ND (Figure **S6** in supporting information). This suggests that there are multiple binding mechanisms between the histag sequence of the protein and QD-ND. The predominant mechanism is likely binding of the histag protein directly to Zn²⁺ ions on the ZnS shell of the QD, owing to the highly compact nature of the peptide coating. Coincidently, this is the route of histag binding to QDs with similarly compact DHLA coatings^{23, 45}. As suggested by our QCM surface assay, a secondary histag binding mechanism can occur through the aspartic acid chelated to Ni²⁺ ions, which may enhance the affinity of QD-ND to the histag.

Figure 4

To validate the use of the QD-ND in a biological system, we chose to probe the efficiency of our protein targeting system with living HeLa cells expressing a histag protein. As a first step, we designed an assay to determine the cytotoxicty of QD-ND. We used the colorimetric MTT assay to monitor HeLa cell viability as a function of both the QD-ND concentration and the QD-ND incubation time (Figure S7 in supporting information). No significant effect on cell viability was observed after one hour of incubation in presence of QD-ND for concentrations ranging from 1 to 100 nM. A decrease in cell viability was seen only for incubation times exceeding 24 hours. In this case, relative cell viability decreased to 60% and 40% for QD-ND concentrations of 75 nM and 100 nM, respectively. Similar results were observed when the QD-ND were preloaded with 50μM NiCl₂ before incubation in presence of the cells (see Figure S8 in supporting information). Based on these observations, QD-ND can be used for cellular labeling in nanomolar concentrations and for incubation periods up to one hour without affecting the cell viability.

Then, we investigated if QD-ND can be used to target a specific protein on the membrane of the HeLa cells. HeLa cells were first transfected with the gene expressing the membrane protein H10-CFP-TM, having the transmembranar domain of platelet-derived growth factor receptor. The extracellular domain of the H10-CFP-TM includes a histag and a cyan fluorescent protein (CFP), which is used to identify cells expressing the protein. To target the H10-CFP-TM, 20 nM QD-ND was incubated with the cells for 10 minutes following by extensive washing with fresh medium. The cells were then examined by fluorescence microscopy. In the CFP channel, approximately 50% of the cells could be identified as expressing H10-CFP-TM (Figure 5a), while cells not expressing the protein were only visible under bright field illumination (Figure 5b). In the QD-ND channel, we observed QD-ND to be colocalized only on cells positive for H10-CFP-TM (Figure 5c and d). The diffusive

movements of individual QD-ND bound to the H10-CFP-TM could be observed as shown in S.Movie1 (see supporting information). In addition, the QD-ND labeling of H10-CFP-TM was eliminated after the addition of free imidazole, a competitor of the histag (Figure **5e**). Combined, these observations demonstrate that the interaction between the cells and QD-ND occurs via the specific histag sequence of the H10-CFP-TM protein. The same experiment performed in the absence of Ni²⁺ loading also led to specific labeling of the H10-CFP-TM protein with QD-ND (Figure **5f-i**), reinforcing our *in vitro* results that binding of histag proteins occurs predominantly at the Zn²⁺ surface of the QD.

Figure 5

To demonstrate that EB1 coupled to QD-ND remains biologically active, we monitored the interaction of the QD-ND-EB1 complex with microtubules during mitotic spindle formation. QD-ND-EB1 was prepared by mixing histag EB1 and QD-ND (using QD with a emission at 545 nm) at a ratio of 6:1. We used *Xenopus* cell extracts, a system known to recapitulate key events of the cell cycle³⁸ and well-suited to monitor microtubule morphogenesis dynamics. The absence of membrane boundaries in the cell extracts allows for direct delivery of the QD-ND-EB1 in the cytosol without requiring endocytosis or micro-injection. Rhodamine-labeled tubulin was initially supplemented to cell extracts to visualize microtubule filaments. The mitotic spindle self-organized to reach a steady-state structure within 30 to 45 minutes and the QD-ND-EB1 complex was applied 15 minutes after the initiation of the assembly. First, we performed static time-point measurements of samples at 30 minutes and 45 minutes. We found that microtubule bundles and QD-ND-EB1 co-localized along microtubule based-structures (Figure 6a-c). The interactions between QD-ND-EB1 and microtubules were specific, as no co-localization between QDs and microtubule structures was observed when cell extracts were similarly treated with QD-ND lacking EB1 or QD-ND complexed with

histag maltose binding protein (MBP), a protein which does not associate with microtubules (Figures **S9a-b** and **S9c-d** in supporting information).

Next, time-lapse experiments were performed in order to monitor the QD-ND-EB1 dynamics within the microtubule-based structures. As shown in S.Movie 2 (see supporting information), most of the QD-ND-EB1 complexes that co-localize with the microtubule-based structures are dynamic. In Figure **6b** and in S.Movie 3 (see supporting information), we report a time sequence of an individual QD-ND-EB1 complex performing a run of 13 µm within 30 seconds along a growing microtubule filament. In another example highlighted in Figure 6c, the QD-ND-EB1 dynamics switched between fast motion and slower motion regimes. As EB1 binds preferentially to the plus-end of microtubules, ³⁷ the QD-ND-EB1 motion we observed reflects the dynamics of microtubule growth in the mitotic spindle.

Figure 6

The velocity histogram of individual events illustrates the heterogeneity of the dynamic QD-ND-EB1 behavior within the spindle structures (Figure 6d). Most dynamic events exhibit a mean velocity within the window of 2 and 5 μm/min. Recently, single molecule experiments performed with the yeast homolog of EB1 in a reconstituted system reported velocities between 2.5 to 10 μm/min ⁴⁶, and tracking of clusters of EB1-alexa594 within mitotic spindles reported an average velocity of 11 μm/min. ⁴⁷ This variability observed in QD-ND-EB1 behavior might correlate with the dynamic properties of the mitotic spindle apparatus. For example, microtubules undergo rapid assembly, disassembly and constant transport, providing multiple mechanisms to influence the dynamics. Both growing microtubules and rapid treadmilling of the tubulin dimers within microtubule filaments drive the observed QD-ND-EB1 dynamics. In addition, processive molecular motors may transport the QD-ND-EB1 complexes along microtubule filaments. These different modes of motion all reflect the complex and heterogeneous nature of the mitotic spindle assembly and maintenance cycles.

CONCLUSION

PEGylated peptides were found to stabilize CdSe/ZnS QDs in aqueous solution and to facilitate direct conjugation of polyhistidine-appended proteins with a controlled stoichiometry. This commercially available peptide offers an alternative way to label and target proteins with QD probes and can be extended to gold nanoparticles using the same functionalization methods. The obtained QD-ND-protein complexes were found to be very stable and of small size, which is highly desirable for cellular labeling. Additional advantages of this labeling approach are the simplicity of the conjugation to histag proteins, the control over the stoichiometry and the reversibility of the binding in presence of competitors. One potential drawback of the method is the requirement of genetic engineering to introduce a polyhistidine sequence into the protein of interest. However, this has become routine practice in most laboratories, as the histag is now a standard method for protein purification.

Our QD-ND probe and labeling strategy was successfully applied to two different biological systems. First, we targeted a histag protein tethered to the external membrane of living cells. Then, we spatially and temporally monitored a microtububle-associated protein, EB1, within mitotic spindles assembled in cell extracts. Both systems demonstrate the ability of QD-ND for single molecule tracking over long time scales not accessible by standard organic fluorescent labels. The possibility to target and study a variety of histag proteins, one at a time, should allow us to further understand how complex biological structures emerge from single molecule events and maintain their integrity despite the transient and rapid turnover of their building blocks.

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SUPPORTING INFORMATION: Figure S1: Effect of the pH on the fluorescence emission of the QD-ND. Figure S2 and Table S1: QCM binding experiments with histag EB1 and peptide ND. Figure S3: Fluorescence quenching effect of QD-ND due to the presence of different divalent ions; Figure S4: FCS and DLS of QD-ND in the presence and absence of EB1 protein, showing the size increase due to the formation of the QD-ND-protein complex. Figure S5: Gel electrophoresis of QD-ND in the presence of increasing concentrations of histag EB1 and small molar ratio of protein/QD in presence of Nickel. Figure S6: Gel electrophoresis of QD-ND in the presence of increasing concentrations of histag EB1 without pre-loading with Ni²⁺ ions. Figures S7 and S8: Cytotoxicity MTT assay with QD-ND and HeLa cells as a function of the QD concentration and the incubation time with or not Ni²⁺ pre-loading of the QD-ND. Figure S9: Controls for the microtubule spindle experiments. S.Movie 1 showing the dynamics of the QD-ND bound to the histag transmembrane receptor at the surface of HeLa cells. Movies S.Movie 2 and S. Movie 3 illustrating the dynamics of QD-

ND-EB1 complexes interacting with the microtubule spindle in cell extracts. This material is available free of charges via the Internet at http://pubs.acs.org.

FIGURE CAPTIONS

Table 1: Hydrodynamic diameters of QD-ND before and after incubation with 1 and 10 equivalents of histag EB1 protein, as measured by fluorescence correlation spectroscopy (FCS) and dynamic light scattering (DLS).

Figure 1: Synthetic route for the PEGylated peptide ND.

Figure 2: a) Chemical structures of peptide **ND** and peptide **7.** b) (left) DLS volumedistribution and (right) agarose gel electrophoresis of QD-ND before (1) and after (2) incubation with 50μM NiCl₂ and the corresponding absorbance/emission spectra of QD-ND before (solid line) and after incubation with NiCl₂ (dashed line).

Figure 3: Fluorescence (top row) and bright field (bottom row) optical microscopy images of HeLa cells after a 10 minute incubation with 100nM of QD-ND (**a** and **c**) and QD-**7** (**b** and **d**) followed by extensive washing.

Figure **4**: **a**) Separation of QD-ND-EB1 complexes by 0.5% agarose gel electrophoresis. QD-ND loaded with Ni²⁺ was incubated with increasing concentrations of histag EB1 protein (n= 1 equivalent of QD-ND calculated from UV absorbance) for 1 hour prior to loading and running the gel. **b**) Gels of the QD-ND-EB1 complex before and after gel extraction and purification of the 1:1 QD-ND:EB1 stoichiometry.

Figure 5: Fluorescence and bright field optical microscopy images of the HeLa cells expressing a transmembrane receptor incorporating an extracellular histag and CFP (H10-CFP-TM). Cells were incubated for 10 minutes with 20nM of QD-ND either in the presence (top row) or absence (bottom row) of Ni²⁺ loading. a and f) CFP channel (red), showing the presence of cells expressing H10-CFP-TM. b and g) brightfield images of all the cells in the field of view. c and h) QD-ND channel (green), showing the binding of QD-ND to cells

expressing the H10-CFP-TM. **d** and **i**) are the superposition of the CFP and QD channels, showing the colocalization of the CFP and QD-ND signals. **e**) QD-ND channel, after treatment of the cells with 100mM imidazole showing the removal of the QD-ND from the cell surface.

Figure **6**: Specific targeting and tracking of QD-ND-EB1 on spindle structures following incubation of the QDs (λ_{em} = 545 nm) with cell extracts. **a**) A rhodamine-labeled spindle structure (red channel) with QD-ND-EB1 (green channel) and the overlay of the two channels (shown in yellow). **b**) Temporal image sequence (5s/frame) of a single QD-ND-EB1 moving on a microtubule (see arrow). **c**) Temporal sequence of a single QD-ND-EB1 (5s/frame) on a microtubule exhibiting a switch between a fast and slow movement. **d**) Mean velocity histogram extracted from a collection of individual QD-ND-EB1 moving on the spindle structures (3 independent experiments, 3 structures).

REFERENCES

- 1. Marks, K. M.; Nolan, G. P., *Nature Methods* **2006**, 3, (8), 591-596.
- 2. Johnsson, N.; Johnsson, K., *Acs Chemical Biology* **2007**, 2, (1), 31-38.
- 3. Giepmans, B. N. G.; Adams, S. R.; Ellisman, M. H.; Tsien, R. Y., *Science*. **2006** 312, 217-224.
- 4. Miller, L. W.; Cornish, V. W., Current Opinion in Chemical Biology **2005**, 9, (1), 56-61.
- 5. Chen, I.; Ting, A. Y., *Current Opinion in Biotechnology* **2005**, 16, (1), 35-40.
- 6. Bonasio, R.; Carman, C. V.; Kim, E.; Sage, P. T.; Love, K. R.; Mempel, T. R.; Springer, T. A.; von Andrian, U. H., *Proceedings of the National Academy of Sciences* **2007**, 104, (37), 14753-14758.
- 7. Howarth, M.; Takao, K.; Hayashi, Y.; Ting, A. Y., *Proceedings of the National Academy of Sciences* **2005**, 102, (21), 7583-7588.
- 8. Michalet, X.; Pinaud, F. F.; Bentolila, L. A.; Tsay, J. M.; Doose, S.; Li, J. J.; Sundaresan, G.; Wu, A. M.; Gambhir, S. S.; Weiss, S., *Science* **2005**, 307, (5709), 538-544.
- 9. Agrawal, A.; Deo, R.; Wang, G. D.; Wang, M. D.; Nie, S., *Proceedings of the National Academy of Sciences* **2008**, 105, (9), 3298-3303.
- 10. Dahan, M.; Levi, S.; Luccardini, C.; Rostaing, P.; Riveau, B.; Triller, A., *Science* **2003**, 302, (5644), 442-445.
- 11. Susumu, K.; Uyeda, H. T.; Medintz, I. L.; Pons, T.; Delehanty, J. B.; Mattoussi, H., *Journal of the American Chemical Society* **2007**, 129, (45), 13987-13996.
- 12. Nehilla, B. J.; Vu, T. Q.; Desai, T. A., *Journal of Physical Chemistry B* **2005**, 109, (44), 20724-20730.
- 13. Iyer, G.; Michalet, X.; Chang, Y.-P.; Pinaud, F. F.; Matyas, S. E.; Payne, G.; Weiss, S., *Nano Letters* **2008**, 8, (12), 4618-4623.
- 14. Mattoussi, H.; Mauro, J. M.; Goldman, E. R.; Anderson, G. P.; Sundar, V. C.; Mikulec, F. V.; Bawendi, M. G., *Journal of the American Chemical Society* **2000**, 122, (49), 12142-12150.
- 15. Choi, H. S.; Liu, W.; Misra, P.; Tanaka, E.; Zimmer, J. P.; Ipe, B. I.; Bawendi, M. G.; Frangioni, J. V., *Nature Biotechnology* **2007**, 25, (10), 1165-1170.
- 16. Swift, J. L.; Cramb, D. T., *Biophysical Journal* **2008**, 95, (2), 865-876.
- 17. Resch-Genger, U.; Grabolle, M.; Cavaliere-Jaricot, S.; Nitschke, R.; Nann, T., *Nat Meth* **2008**, 5, (9), 763-775.
- 18. Howarth, M.; Liu, W.; Puthenveetil, S.; Zheng, Y.; Marshall, L. F.; Schmidt, M. M.; Wittrup, K. D.; Bawendi, M. G.; Ting, A. Y., *Nature Methods* **2008**, 5, 397 399
- 19. O'Hare, H. M.; Johnsson, K.; Gautier, A., Current Opinion in Structural Biology 2007, 17, (4), 488-494.
- 20. Roullier, V.; Clarke, S.; You, C.; Pinaud, F.; Gouzer, G.; Schaible, D.; Marchi-Artzner, V.; Piehler, J.; Dahan, M., *Nano Letters* **2009**, 9, (3), 1228-1234.
- 21. Tekle, C.; van Deurs, B.; Sandvig, K.; Iversen, T. G., *Nano Letters* **2008**, 8, (7), 1858-1865.
- 22. Groc L; Lafourcade M; Heine M; Renner M; Racine V; Sibarita JB; Lounis B; Choquet D; L., C., *J Neurosci.* **2007 Nov 14**, 27(46), 12433-7
- 23. Sapsford, K. E.; Pons, T.; Medintz, I. L.; Higashiya, S.; Brunel, F. M.; Dawson, P. E.; Mattoussi, H., *Journal of Physical Chemistry C* **2007**, 111, (31), 11528-11538.
- 24. Medintz, I. L.; Berti, L.; Pons, T.; Grimes, A. F.; English, D. S.; Alessandrini, A.; Facci, P.; Mattoussi, H., *Nano Letters* **2007**, 7, 1741-1748.

- 25. Medintz, I. L.; Sapsford, K. E.; Clapp, A. R.; Pons, T.; Higashiya, S.; Welch, J. T.; Mattoussi, H., *Journal of Physical Chemistry B* **2006**, 110, (22), 10683-10690.
- 26. Liu, W.; Howarth, M.; Greytak, A. B.; Zheng, Y.; Nocera, D. G.; Ting, A. Y.; Bawendi, M. G., *J. Am. Chem. Soc.* **2008**, 130, (4), 1274 1284.
- 27. Delehanty, J. B.; Medintz, I. L.; Pons, T.; Brunel, F. M.; Dawson, P. E.; Mattoussi, H., *Bioconjugate Chemistry* **2006**, 17, (4), 920-927.
- 28. Medintz, I. L.; Goldman, E. R.; Lassman, M. E.; Mauro, J. M., *Bioconjugate Chemistry* **2003**, 14, (5), 909-918.
- 29. Pellegrino, T.; Manna, L.; Kudera, S.; Liedl, T.; Koktysh, D.; Rogach, A. L.; Keller, S.; Radler, J.; Natile, G.; Parak, W. J., *Nano Letters* **2004**, 4, (4), 703-707.
- 30. Pons, T.; Uyeda, H. T.; Medintz, I. L.; Mattoussi, H., *The Journal of Physical Chemistry B* **2006**, 110, (41), 20308-20316.
- 31. Carion, O.; Mahler, B.; Pons, T.; Dubertret, B., *Nature Protocols* **2007**, 2, (10), 2383-2390.
- 32. Luccardini, C.; Tribet, C.; Vial, F.; Marchi-Artzner, V.; Dahan, M., *Langmuir* **2006**, 22, (5), 2304-2310.
- 33. Mei, B. C.; Susumu, K.; Medintz, I. L.; Delehanty, J. B.; Mountziaris, T. J.; Mattoussi, H., *Journal of Materials Chemistry* **2008**, 18, (41), 4949-4958.
- 34. Mei, B. C.; Susumu, K.; Medintz, I. L.; Mattoussi, H., *Nat Protoc* **2009**, 4, (3), 412-23.
- 35. Pinaud, F.; King, D.; Moore, H. P.; Weiss, S., *Journal of the American Chemical Society* **2004**, 126, (19), 6115-6123.
- 36. Dif, A.; Henri, E.; Artzner, F.; Baudy-Floch, M.; Schmutz, M.; Dahan, M.; Marchi-Artzner, V., *J. Am. Chem. Soc.* **2008**, 130, (26), 8289-8296.
- 37. Schuyler, S. C.; Pellman, D., *Cell* **2001**, 105, (4), 421-424.
- 38. Desai, A., Methods Cell Biol 1999, 61, 385
- 39. Hyman, A. A., J. Cell. Sci. Suppl. 1991, 14, (125).
- 40. Tirnauer, J. S.; Grego, S.; Salmon, E. D.; Mitchison, T. J., *Mol. Biol. Cell* **2002**, 13, (10), 3614-3626.
- 41. Leatherdale, C. A.; Woo, W. K.; Mikulec, F. V.; Bawendi, M. G., *The Journal of Physical Chemistry B* **2002**, 106, (31), 7619-7622.
- 42. Yu, W. W.; Qu, L.; Guo, W.; Peng, X., Chemistry of Materials **2003**, 15, (14), 2854-2860.
- 43. Marx, K. A., Biomacromolecules **2003**, 4, (5), 1099-1120.
- 44. Höök, F.; Vörös, J.; Rodahl, M.; Kurrat, R.; Boni, P.; Ramsden, J. J.; Textor, M.; Spencer, N. D.; Tengvall, P.; Gold, J.; Kasemo, B., *Colloids Surf.*, B **2002**, 24, 155-170.
- 45. Pons, T.; Medintz, I. L.; Wang, X.; English, D. S.; Mattoussi, H., *Journal of the American Chemical Society* **2006**, 128, (47), 15324-15331.
- 46. Bieling, P.; Laan, L.; Schek, H.; Munteanu, E. L.; Sandblad, L.; Dogterom, M.; Brunner, D.; Surrey, T., *Nature* **2007**, 450, (7172), 1100-1105.
- 47. Tirnauer, J. S.; Salmon, E. D.; Mitchison, T. J., Mol. Biol. Cell **2004**, 15, (4), 1776-1784.

TOC graphics

