



## FTIR spectroscopic study of an organic/mineral composite for bone and dental substitute materials.

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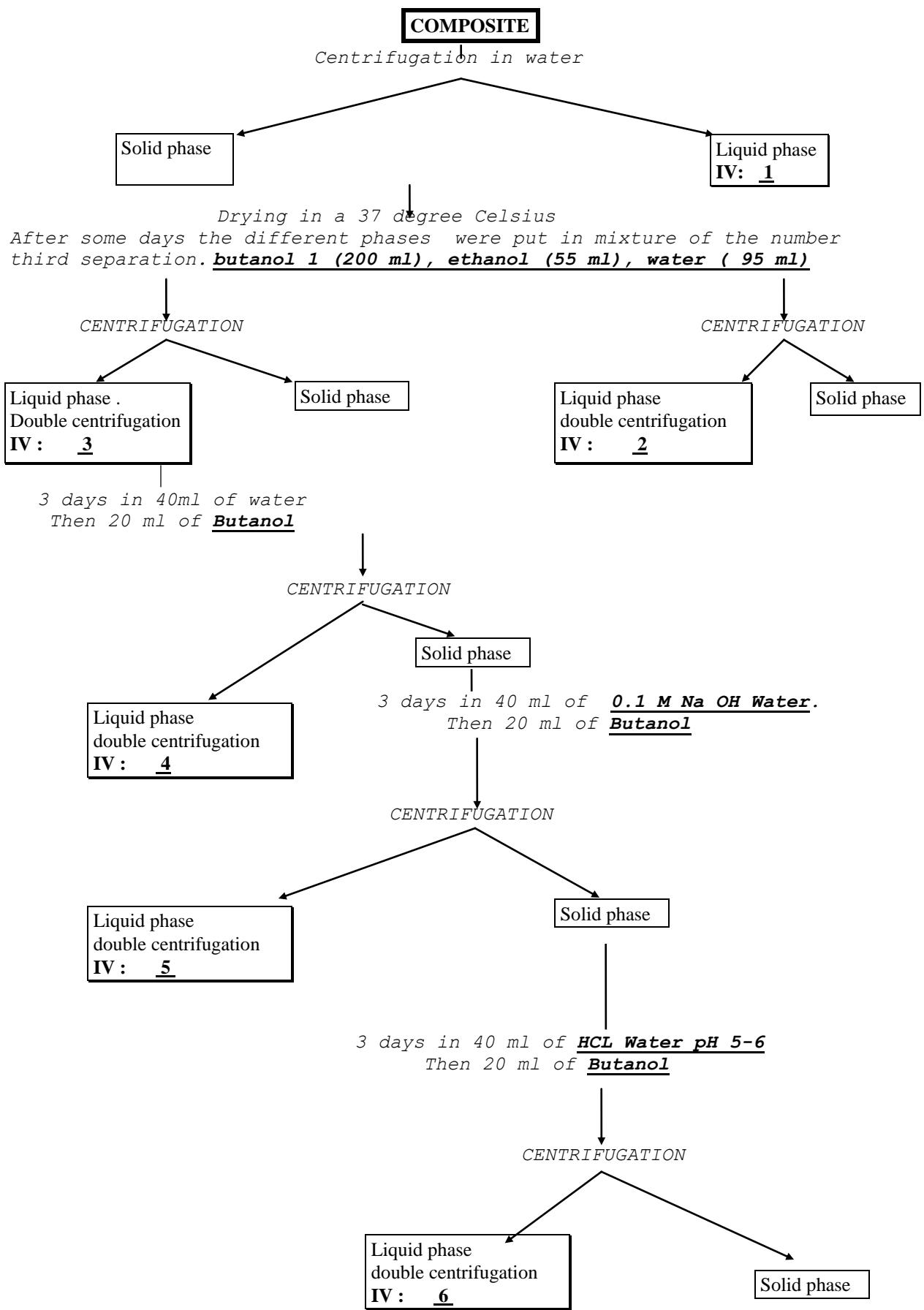
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<i>PEAK</i>	<i>Possible assignments</i>	<i>References</i>
712	Peak of carbonates --> <b>MINERAL</b>	11
870- 874	absorption of C-O in $\text{CO}_3^{2-}$ --> <b>MINERAL</b>	13 -11
1000-1200	C-O alcohol and C-O-C ---> <b>ORGANIC</b>	15 - 14 -19 - 18 - 20
1225-1235	Vib de def plane de OH et <b>COOH</b>	15
1250-1500	$\text{CH}_2 \text{ CH}_3$ ----> <b>ORGANIC</b>	14 -18 - 16 - 15
1360 - 1450	Weaker symmetric <b>COO</b> stretching vibration	14
1410 - 1420	absorption of C-O in $\text{CO}_3^{2-}$ --> <b>MINERAL</b>	13 - 11 -12
1460	absorption of C-O in $\text{CO}_3^{2-}$ --> <b>MINERAL</b>	13 - 12
1540 - 1650	Strong asymmetric <b>COO<sup>-</sup></b> stretching vibration	14
1550 -1900	<b>C=O</b>	14
1563 - 1667	Stretching vibrations of the <b>COO<sup>-</sup></b>	20
1592	<b>COO<sup>-</sup></b> Salt formation	20
1667-1724	Carbonyl <b>C=O</b> Oxidation	20
1724 - 2000	<b>C=O</b> stretching	20 - 15
1795	small peak of carbonates --> <b>MINERAL</b>	11
1754	<b>C=O</b> monosaccharides	18
2700 - 3000	$\text{CH} \text{ CH}_2 \text{ CH}_3$ ----> <b>ORGANIC</b>	17- 15 - 14 - 18

<b>Minerals fillers</b>	<b>pH</b>
BCP 1	11
BCP 1 W	9-10
BCP 2	6
HA	8
OHAP	10

pH/medium	T0	T1	proton consumption $\text{mol} \cdot \text{l}^{-1}$
water	9,03	9,00	-6,64E-11
BCP 1g/20ml	8,77	10,62	1,68E-09
Ca(OH)2 0.1M	11,96	12,40	6,86E-13
NaOH 0.1M	11,62	12,48	2,06E-12
NaOH 1M	13,00	13,17	3,27E-14

Presence of	<b>T0</b>			<b>T1</b>			<b>T2</b>		
	<u>COO<sup>-</sup></u>	<u>C=O</u>	<u>Mineral*</u>	<u>COO<sup>-</sup></u>	<u>C=O</u>	<u>Mineral*</u>	<u>COO<sup>-</sup></u>	<u>C=O</u>	<u>Mineral*</u>
<b>BCP</b>	No	No	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>	No	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>
<b>HA</b>				No	<b>Yes</b>	No	No	<b>Yes</b>	No
<b>OHAP</b>	No	No	No	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>	<b>Yes</b>

\* CaCO<sub>3</sub>

separation number:		<b>IV- 1</b>	<b>IV- 2</b>	<b>IV- 3</b>	<b>IV- 4</b>	<b>IV- 5</b>	<b>IV- 6</b>	
FILLERS	TIMES	Water	Mixture of alcohols + water	Mixture of alcohols + water	Butanol + water	Butanol + alkalin solution	Butanol + acidic solution	<b>TOTAL</b>
<b>BCP 1</b> pH :11	T0 T3	Yes		Yes No	Yes Yes	Yes Yes	No Yes	<b>4/5</b> <b>4/5</b>
<b>BCP 1 W</b> pH : 9-10	T2		Yes	Yes	Yes	Yes	No	<b>4/5</b>
<b>BCP 2</b> pH : 6	T1		No	No	No	Yes	No	<b>1/5</b>
<b>HA</b> pH : 8	T3		No	No		Yes	No	<b>1/5</b>
<b>OHAP</b> pH : 10	T0 T3	Yes		Yes No	Yes Yes	Yes Yes	No Yes	<b>4/5</b> <b>3/5</b>
<b>TOTAL</b>		<b>2/2</b>	<b>2/5</b>	<b>3/7</b>	<b>5/6</b>	<b>7/7</b>	<b>2/7</b>	

**PRESENCE OF COO PEAKS**

	T0	T1	T2	T3
BCP 1	0			1
BCP 1 W			0	
BCP 2		0		
HA				1
OHAP	0			3

**PRESENCE OF C=O PEAKS**

	T0	T1	T2	T3
BCP 1	1			2
BCP 1 W			3	
BCP 2		4		
HA				2
OHAP	4			4